

Crystal and Molecular Structure of Di- μ -(tricarbonyl- π -cyclopentadienyl-tungsten-OO')bisdimethylaluminium

By G. J. GAINSFORD, R. R. SCHRIEKE, and J. D. SMITH*

(School of Molecular Sciences, University of Sussex, Brighton BN1 9QJ)

Summary The compound $[(C_5H_5)W(CO)_3AlMe_2]_2$ consists of molecules having twelve-membered rings with Al-O-C-W bridges between aluminium and tungsten.

In many molecules, carbonyl groups form bridges by coordination of carbon to two or more metal atoms. We now describe a carbonyl bridge in which carbon is linked to one metal atom and oxygen to another.

Crystals of the compound $[(C_5H_5)W(CO)_3AlMe_2]_2$,¹ which is exceedingly reactive towards traces of air or moisture, were shaken from the reaction vessel, and sealed in evacuated thin-walled Pyrex capillaries. They were shown by X-ray methods to be monoclinic with $a = 18.120(6)$, $b = 6.188(3)$, $c = 22.266(8)$ Å, $\beta = 93.29(2)^\circ$; $Z = 8$ $\{[(C_5H_5)W(CO)_3AlMe_2]$ units}; space group $C2/c$. Intensity data for 1600 reflections with $F^2_{obs}/\sigma(F^2_{obs}) \geq 3.0$ were observed

by four-circle automatic diffractometry using Mo- K_{α} radiation, and atomic parameters were found by Patterson, Fourier, and least-squares methods. All atoms were refined with anisotropic thermal parameters giving R 0.07 and further refinement with data corrected for absorption (μ 98.7 cm^{-1}) is proposed. The analysis shows well separated centrosymmetric dimeric molecules (Figure). Each tungsten is bound to a cyclopentadienyl group and to three carbonyl groups. The oxygen atoms of two of the

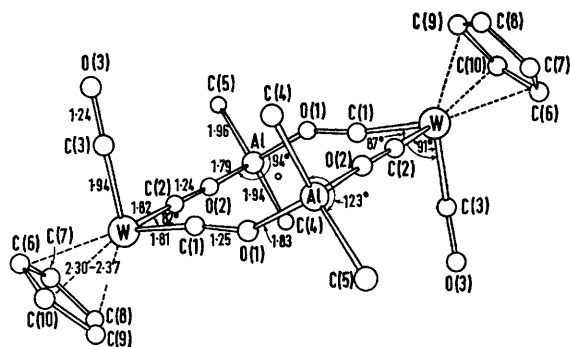


FIGURE. Distances are in Å.

carbonyls are linked to aluminium atoms making a slightly puckered twelve-membered ring. The tungsten achieves roughly octahedral co-ordination (with the cyclopentadienyl group occupying three co-ordination sites) and the aluminium is four-co-ordinate with the wide exocyclic C-Al-C angle found in other derivatives $(\text{Me}_2\text{AlX})_n$ ($n = 2, 3$; X = e.g., Cl, Me, NR_2).² The nucleophile $[(\text{C}_5\text{H}_5)\text{W}(\text{CO})_3]^-$ may thus serve as a bridging group between aluminium atoms in this class of compounds.

Estimated standard deviations calculated from the inverse least-squares matrix are 0.03 (W-C, Al-C, Al-O) and 0.045 Å (C-C, C-O). Distances are uncorrected for thermal motion. The W-C bond lengths within the ring appear to be short compared with tungsten-terminal carbonyl bonds.³ All C-O distances appear to be long compared with those in other carbonyl compounds, but discussion of molecular parameters is deferred until refinement is complete.

Similar carbonyl bridges between tungsten and aluminium in the ion $[(\text{C}_5\text{H}_5)\text{W}(\text{CO})_3\text{AlPh}_3]^-$ were suggested⁴ by i.r. spectroscopic data.

We thank the S.R.C. for financial support.

(Received, 7th April 1972; Com. 592.)

¹ R. R. Schrieke and J. D. Smith, *J. Organometallic Chem.*, 1971, **31**, C46.

² G. Allegra, G. Perego, and A. Immirzi, *Macromol. Chem.*, 1963, **61**, 69; J. C. Huffman and W. E. Streib, *Chem. Comm.*, 1971, 911; K. Gosling, G. M. McLaughlin, G. A. Sim, and J. D. Smith, *ibid.*, 1970, 1617; H. Hess, A. Hinderer, and S. Steinhauser, *Z. anorg. Chem.*, 1970, **377**, 1.

³ V. A. Semion and Yu. T. Struchkov, *J. Struct. Chem.*, 1968, **9**, 931.

⁴ J. M. Burlitch and R. B. Petersen, *J. Organometallic Chem.*, 1970, **24**, C65.